

FORM PTO-1390  
(REV. 10-96)

U.S. DEPARTMENT OF COMMERCE PATENT AND TRADEMARK OFFICE

ATTORNEY DOCKET NUMBER

A-71327/DJB/MAK

**TRANSMITTAL LETTER TO THE UNITED STATES  
DESIGNATED/ELECTED OFFICE (DO/EO/US)  
CONCERNING A FILING UNDER 35 U.S.C. 371**

U.S. APPLICATION NO. (If known, sect. 37 C.F.R. 1.5)

10/049705  
Not Yet Known

INTERNATIONAL APPLICATION NO.

PCT/AU00/00974

INTERNATIONAL FILING DATE

16 August 2000

PRIORITY DATE CLAIMED

16 August 1999

TITLE OF INVENTION

FUEL CELL SYSTEM

APPLICANT(S) FOR DO/EO/US

Karl FOGER and Khaliq AHMED

Applicant herewith submits to the United States Designated/Elected Office (DO/EO/US) the following items and other information:

1. ☒ This is a FIRST submission of items concerning a filing under 35 U.S.C. 371.
2. ☐ This is a SECOND or SUBSEQUENT submission of items concerning a filing under 35 U.S.C. 371
3. ☒ This is an express request to begin national examination procedures (35 U.S.C. 371(f)) at any time rather than delay examination until the expiration of the applicable time limit set in 35 U.S.C. 371(b) and PCT Articles 22 and 39(1).
4. ☐ A proper Demand for International Preliminary Examination was made by the 19th month from the earliest claimed priority date.
5. ☒ A copy of the International Application as filed (35 U.S.C. 371(c)(2))
  - a. ☐ is transmitted herewith (required only if not transmitted by the International Bureau).
  - b. ☒ has been transmitted by the International Bureau. (see enclosed Form PCT/IB/308)
  - c. ☐ is not required, as the application was filed in the United States Receiving Office (RO/US)
6. ☐ A translation of the International Application into English (35 U.S.C. 371(c)(2)).
7. ☒ Amendments to the claims of the International Application under PCT Article 19 (35 U.S.C. 371(c)(3))
  - a. ☐ are transmitted herewith (required only if not transmitted by the International Bureau).
  - b. ☒ have been transmitted by the International Bureau.
  - c. ☐ have not been made; however, the time limit for making such amendments has NOT expired.
  - d. ☐ have not been made and will not be made.
8. ☐ A translation of the amendments to the claims under PCT Article 19 (35 U.S.C. 371(c)(3)).
9. ☐ An oath or declaration of the inventor(s) (35 U.S.C. 371(c)(4)).
10. ☐ A translation of the annexes to the International Preliminary Examination Report under PCT Article 36 (35 U.S.C. 371(c)(5)).

Items 11. to 16. below concern other document(s) or information included:

11. ☐ An Information Disclosure Statement under 37 CFR 1.97 and 1.98.
12. ☐ An assignment document for recording. A separate cover sheet in compliance with 37 CFR 3.28 and 3.31 is included.
13. ☒ A FIRST preliminary amendment.
  - ☐ A SECOND or SUBSEQUENT preliminary amendment.
14. ☐ A substitute specification.
15. ☐ A change of power of attorney and/or address letter.
16. ☐ Other items or information.

U.S. APPLICATION NO. (If known, see 37 C.F.R. 1.53) Not Yet Known	INTERNATIONAL APPLICATION NO. <u>10/049705</u> PCT/AU00/00974	ATTORNEY'S DOCKET NUMBER A-71327/DJB/MAK
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17. <input checked="" type="checkbox"/> The following fees are submitted:					<b>CALCULATIONS (PTO USE ONLY)</b>	
<b>Basic National Fee (37 CFR 1.492(a)(1)-(5)):</b>						
Search Report has been prepared by the EPO or JPO ..... \$890.00						
International preliminary examination fee paid to USPTO (37 CFR 1.482) ..... \$710.00						
No international preliminary examination fee paid to USPTO (37 CFR 1.482) but international search fee paid to USPTO (37 CFR 1.445(a)(2)) ..... \$740.00						
Neither international preliminary examination fee (37 CFR 1.482) nor international search fee (37 CFR 1.445(a)(2)) paid to USPTO ..... \$1,040.00						
International preliminary examination fee paid to USPTO (37 CFR 1.482) and all claims satisfied provisions of PCT Article 33(1)-(4) ..... \$100.00						
<b>ENTER APPROPRIATE BASIC FEE AMOUNT =</b>					\$	1,040.00
Surcharge of \$130.00 for furnishing the oath or declaration later than months from the earliest claimed priority date (37 CFR 1.492(e)). <input type="checkbox"/> 20 <input checked="" type="checkbox"/> 30					\$	130.00
<b>CLAIMS</b>	<b>NUMBER FILED</b>	<b>NUMBER EXTRA</b>	<b>RATE</b>			
Total Claims	21	-20 =	1	\$ 18.00	\$	18.00
Independent Claims	1	-3 =	0	\$ 84.00	\$	0.00
Multiple dependent claims (if applicable)				\$280.00	\$	0.00
<b>TOTAL OF ABOVE CALCULATIONS =</b>					\$	1,188.00
Reduction by 1/2 for filing by small entity, if applicable. Applicant claims small entity status. (See 37 CFR 1.27.)					Yes	No
						✓
<b>SUBTOTAL =</b>					\$	1,188.00
Processing fee of \$130.00 for furnishing the English translation later than months from the earliest claimed priority date (37 CFR 1.492(f)). <input type="checkbox"/> 20 <input type="checkbox"/> 30					\$	0.00
<b>TOTAL NATIONAL FEE =</b>					\$	1,188.00
Fee for recording the enclosed assignment (37 CFR 1.21(h)). The assignment must be accompanied by an appropriate cover sheet (37 CFR 3.28, 3.31). \$40.00 per property					\$	0.00
<b>TOTAL FEES ENCLOSED =</b>						1,188.00
					Amount to be refunded	\$
					charged	\$
a. <input checked="" type="checkbox"/> A check in the amount of <u>\$1,188.00</u> to cover the above fees is enclosed.						
b. <input type="checkbox"/> Please charge my Deposit Account No. 06-1300 (Order No. _____) in the amount of \$ _____ to cover the above fees.						
c. <input checked="" type="checkbox"/> The Commissioner is hereby authorized to charge any additional fees which may be required, or credit any overpayment to Deposit Account No. 06-1300 (Order No. <u>A-71327/DJB/MAK</u> ).						
<b>NOTE:</b> Where an appropriate time limit under 37 CFR 1.494 or 1.495 has not been met, a petition to revive (37 CFR 1.137(a) or (b)) must be filed and granted to restore the application to pending status.						
<b>SEND ALL CORRESPONDENCE TO:</b>						
Michael A. KAUFMAN FLEHR HOHBACH TEST ALBRITTON & HERBERT LLP Four Embarcadero Center - Suite 3400 San Francisco, California 94111-4187 Tel.: (415) 781-1989 Fax: (415) 398-3249						
					<i>Michael A. Kaufman</i> SIGNATURE	
					Michael A. KAUFMAN NAME	
					32,998 REGISTRATION NUMBER	

**PATENT**

Attorney Docket No.: A-71327/DJB/MAK

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re application of:

Karl FOGER and Khaliq AHMED

**NATIONAL PHASE ENTRY OF:**

PCT/AU00/00974

Serial No.: Not Yet Assigned

Filing Date: Herewith

For: *Fuel Cell System*

Box PCT APPLICATION  
Assistant Commissioner for Patents  
Washington, D.C. 20231

FIRST PRELIMINARY AMENDMENT

Sir:

This Preliminary Amendment accompanies the filing of an application under 35 U.S.C. § 371.

Prior to examination, please amend the above-identified application as follows:

IN THE SPECIFICATION:

Please replace the paragraph beginning on page 2 at line 22 with the following rewritten paragraph:

--One proposal of a fuel cell electricity generation process in which a hydrocarbon fuel is converted to a fuel cell fuel stream including hydrogen in a steam pre-reformer is disclosed in EP-A-0435724. The temperature in the pre-reformer is



6. (Amended) A process according to claim 1 in which the steam to carbon ratio in the methane generator is no more than 1.5.
9. (Amended) A process according to claim 1 in which the fuel is a C<sub>3+</sub> hydrocarbon fuel.
11. (Amended) A process according to claim 1 in which the fuel is selected from the group consisting of ethane, propane, butane, LPG, gasoline (petrol), diesel, kerosene, fuel oil, jet oil, naphtha and mixtures of two or more of these.
12. (Amended) A process according to claim 1 in which the reaction at the anode of the fuel cell is performed at a temperature of at least 700°C.
13. (Amended) A process according to claim 1 in which waste heat from the fuel cell is recycled to the steam pre-reformer.
14. (Amended) A process according to claim 1 in which the reaction in the steam pre-reformer results in the conversion of at least 97.5% of the higher carbon (C<sub>2+</sub>) hydrocarbon fuel.

--16. A process according to claim 1 in which the fuel stream includes no less than about 50% by volume methane measured on a wet basis.

17. A process according to claim 1 in which the fuel stream includes no less than about 60% by volume methane measured on a wet basis.
18. A process according to claim 1 in which the temperature in the steam pre-reformer is in a range of about 250°C to 450°C.

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**U.S. Filing Date:** Herewith

19. A process according to claim 1 in which the temperature in the steam pre-reformer is in a range of about 300°C to 400°C.

20. A process according to claim 6 in which the steam to carbon ratio in the methane generator is no more than 1.25.

21. A process according to claim 6 in which the steam to carbon ratio in the methane generator is no more than 1.0.--

IN THE ABSTRACT OF THE DISCLOSURE:

Please add the following new paragraph and heading at the top of a new page immediately following claim 21:

--ABSTRACT OF THE DISCLOSURE

A fuel cell produces electricity by reacting a higher carbon hydrocarbon fuel with steam in a steam pre-reformer, whose temperature does not exceed 500°C. A fuel stream is produced that includes hydrogen and not less than about 20% by volume methane, measured on a wet basis. The fuel stream and an oxidant are supplied to a high temperature fuel cell in which the methane is reformed. The fuel cell produces electricity by reacting the fuel stream at a fuel cell anode, and by reacting the oxidant at a fuel cell cathode.--

REMARKS

The foregoing claim amendments are made to obviate the problem of improper dependency in the multiple dependent claims which existed in the PCT priority document. Also, new claims 16-21 add no new material as they present language (now deleted) from existing claims as previously filed. The two amendments to the specification are to correct obvious typographical errors. The first error was a simple transposition and the second error omitted a word clearly contained in claim 11 as originally filed. The foregoing abstract of the disclosure

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was added because no abstract appeared in the PCT priority document. Appended hereto at page 4 is a marked-up version of the foregoing amendments in which additions to the text are shown with a gray background and deletions with .

The Commissioner is authorized to charge any additional fees which may be required, or credit any overpayment to Deposit Account No. 06-1300 (Our Order No. A-71327/DJB/MAK).

Respectfully submitted,

FLEHR, HOHBACH, TEST,  
ALBRITTON & HERBERT LLP

Dated: 14 February 2002

By: Michael A. Kaufman  
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SF-10705089v1

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**U.S. Filing Date:** Herewith

VERSION WITH MARKINGS TO SHOW CHANGES MADE

IN THE SPECIFICATION:

The paragraph beginning on page 2 at line 22 has been amended as follows:

One proposal of a fuel cell electricity generation process in which a hydrocarbon fuel is converted to a fuel cell fuel stream including hydrogen in a steam pre-reformer is disclosed in EP-A-0435724. The temperature in the pre-reformer is described as 700 to 850°C with a resultant product-gas composition of 65-80 vol% H<sub>2</sub>, ~~2-50~~ <sup>5-20</sup> vol% CO, and 5-25 vol% CO<sub>2</sub>.

The paragraph beginning at page 4, line 16 has been amended as follows:

By the present invention, a substantially wider source of fuel may be used for the fuel cell than just methane and/or hydrogen, including ethane and liquid higher hydrocarbons such as propane, butane, liquefied petroleum gas (LPG), gasoline (petrol), diesel, kerosene, fuel oil, jet oil, <sup>and</sup> ~~naphtha~~ <sup>naphtha</sup> and mixtures of these, while a lower temperature of no greater than 500°C may be used for steam pre-reforming the higher hydrocarbon fuel source since there is no requirement to reform any methane in the steam pre-reformer. This permits a relatively small pre-reformer reactor to be used which, combined with the reduced maximum operating temperature of 500°C, enables a simplified and therefore cheaper pre-reformer system to be adopted. Such low temp steam pre-reforming also alleviates carbon deposition in the pre-reformer.

IN THE CLAIMS:

Claims 3, 4, 5, 6, 9, 11, 12, 13 and 14 have been amended as follows:

3. (Amended) A process according to claim ~~2~~ <sup>1</sup> in which the fuel stream includes no less than about 40% by volume methane, ~~even more preferably no less than about 50% by volume methane, and most preferably no less than about 60% by volume methane,~~ measured on a wet basis.



$\frac{d}{dt} \left( \frac{\partial L}{\partial \dot{x}} \right) = \frac{\partial L}{\partial x}$

- New claims 16 through 21 have been added as follows:

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17. A process according to claim 1 in which the fuel stream includes no less than about 60% by volume methane measured on a wet basis.
18. A process according to claim 1 in which the temperature in the steam pre-reformer is in a range of about 250°C to 450°C.
19. A process according to claim 1 in which the temperature in the steam pre-reformer is in a range of about 300°C to 400 °C.
20. A process according to claim 6 in which the steam to carbon ratio in the methane generator is no more than 1.25.
21. A process according to claim 6 in which the steam to carbon ratio in the methane generator is no more than 1.0.

IN THE ABSTRACT OF THE DISCLOSURE:

The following text has been added as an abstract of the disclosure:

ABSTRACT OF THE DISCLOSURE

A fuel cell produces electricity by reacting a higher carbon hydrocarbon fuel with steam in a steam pre-reformer, whose temperature does not exceed 500°C. A fuel stream is produced that includes hydrogen and not less than about 20% by volume methane, measured on a wet basis. The fuel stream and an oxidant are supplied to a high temperature fuel cell in which the methane is reformed. The fuel cell produces electricity by reacting the fuel stream at a fuel cell anode, and by reacting the oxidant at a fuel cell cathode.

## FUEL CELL SYSTEM

### Technical Field

5

The present invention relates to fuel cells and is particularly concerned with the fuel supply to a fuel cell electrical power generating system.

### Background Art

10

Fuel cells convert gaseous fuels (such as hydrogen, natural gas and gasified coal) via an electrochemical process directly into electricity. A fuel cell continuously produces power when supplied with fuel and oxidant, normally air. A typical fuel cell consists of an electrolyte (ionic conductor,  $H^+$ ,  $O^{2-}$ ,  $CO_3^{2-}$  etc.) in contact with two electrodes (mainly electronic conductors).

15

On shorting the cell through an external load, fuel oxidises at the anode resulting in the release of electrons which flow through the external load and reduce oxygen at the cathode. The charge flow in the external circuit is balanced by ionic current flows within the electrolyte. Thus, at the cathode oxygen from the air or other oxidant is dissociated and converted to oxygen ions which migrate through the electrolyte membrane and react with the fuel at the anode/electrolyte interface. The voltage from a single cell under load conditions is in the vicinity of 0.6 to 1.0 V DC and current densities in the range 100 to 1000  $mAcm^{-2}$  can be achieved.

20

Several different types of fuel cells have been proposed. Amongst these, the solid oxide fuel cell (SOFC) is regarded as the most efficient and versatile power generation system, in particular for dispersed power generation, with low pollution, high efficiency, high power density and fuel flexibility. SOFC's operate at elevated temperatures, for example 700 - 1000°C. Other fuel cells which operate at elevated temperatures include the molten carbonate fuel cell requiring a minimum temperature of 650°C. However, SOFC's are the primary interest for the invention and further discussion herein will be mainly directed to these without intending to be limited in any way.

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- 2 -

Numerous SOFC configurations are under development, including the tubular, the monolithic and the planar design. The planar or flat plate design is the most widely investigated. Single planar SOFC's are connected via interconnects or gas separators to form multi-cell units, sometimes termed fuel cell stacks. Gas flow paths are provided between the gas separators and  
5 respective electrodes, for example by providing gas flow channels in the gas separators. In a fuel cell stack the components - electrolyte/electrode laminates and gas separator plates are fabricated individually and then stacked together. With this arrangement, external and internal co-flow, counter-flow and cross-flow manifolding options are possible for the gaseous fuel and oxidant.

10

Traditionally hydrogen, usually moistened with steam, has been used as a fuel cell fuel. However, in order to be economically viable the fuel must be as cheap as possible. One relatively cheap source of hydrogen is natural gas, primarily methane with a small proportion of heavy hydrocarbons ( $C_2+$ ). Natural gas is commonly converted to hydrogen in a steam  
15 reforming reaction, but the reaction is endothermic and, because of the stability of methane, requires a reforming temperature of at least about  $650^{\circ}C$  for substantial conversion and a higher temperature for complete conversion. While high temperature fuel cell systems produce heat which must be removed, heat exchangers capable of transferring thermal energy at the required level of at least about  $650^{\circ}C$  from the fuel cells to a steam reformer are expensive.  
20 Thus, hydrogen produced by steam reforming natural gas may not be a cheap source of fuel.

One proposal of a fuel cell electricity generation process in which a hydrocarbon fuel is converted to a fuel cell fuel stream including hydrogen in a steam pre-reformer is disclosed in EP-A-0435724. The temperature in the pre-reformer is described as 700 to  $850^{\circ}C$  with a  
25 resultant product-gas composition of 65-80 vol%  $H_2$ , 2-50 vol% CO, and 5-25 vol%  $CO_2$ .

Another such proposal is disclosed in US-A- 5,302,470 in which the steam pre-reforming reaction is said to be carried out under similar conditions to those of known steam reforming reactions: for example, an inlet temperature of about 450 to  $650^{\circ}C$ , an outlet temperature of  
30 about 650 to  $900^{\circ}C$ , and a pressure of about 0 to 10 kg/cm<sup>2</sup>.G to produce a fuel cell fuel stream which is composed mainly of hydrogen and is fed to the fuel cell anode via a carbon monoxide

Indirect internal steam reforming of the remaining hydrocarbon fuel within the fuel inlet passages is said to allow the use of reforming catalysts within the fuel inlet passages which are less likely to produce coking or carbon deposits from the internal steam reforming of the

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higher hydrocarbons than nickel cermet anodes. It is believed that steam pre-reforming of the hydrocarbon fuel in the described temperature range is restricted to 5 to 20% of the fuel in order to relatively increase the level of hydrogen in the fuel stream to the fuel cell stack and thereby alleviate carbon deposition when the fuel is internally reformed in the stack.

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### Summary of the Invention

According to the present invention there is provided a process for producing electricity in a fuel cell which comprises reacting a higher carbon ( $C_{2+}$ ) hydrocarbon fuel with steam in a steam pre-reformer at a temperature in the pre-reformer of no greater than  $500^{\circ}\text{C}$  to produce a fuel stream including hydrogen and no less than about 20% by volume methane measured on a wet basis, and supplying the fuel stream and an oxidant to a high temperature fuel cell in which the methane is reformed and electricity is produced by reacting the fuel stream at an anode of the fuel cell and reacting the oxidant at a cathode of the fuel cell.

15

By the present invention, a substantially wider source of fuel may be used for the fuel cell than just methane and/or hydrogen, including ethane and liquid higher hydrocarbons such as propane, butane, liquefied petroleum gas (LPG), gasoline (petrol), diesel, kerosene, fuel oil, jet oil and mixtures of these, while a lower temperature of no greater than  $500^{\circ}\text{C}$  may be used for steam pre-reforming the higher hydrocarbon fuel source since there is no requirement to reform any methane in the steam pre-reformer. This permits a relatively small pre-reformer reactor to be used which, combined with the reduced maximum operating temperature of  $500^{\circ}\text{C}$ , enables a simplified and therefore cheaper pre-reformer system to be adopted. Such low temp steam pre-reforming also alleviates carbon deposition in the pre-reformer.

25

Steam pre-reforming is conveniently performed at atmospheric pressure, but higher pressures may be adopted if desired, for example upto  $10\text{ kgcm}^{-2}\text{G}$ .

The fuel source may include non-higher hydrocarbons, such as methane, but preferably the higher hydrocarbons form the major component of the fuel source. The preferred fuel is selected from LPG, gasoline (petrol) and diesel.

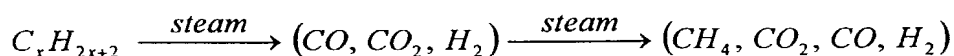
30

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Steam pre-reforming of the higher hydrocarbon fuel source is preferably performed at a temperature no greater than about 450°C, more preferably in a range of about 250 to 450°C, and, depending upon the fuel and other process parameters, most preferably in a range of about 300 to 400°C.

5

The relatively low maximum temperature in the steam pre-reformer tends to favour methane formation over the reforming catalyst in the reactions:



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Thus, methane is formed by steam reforming of the higher hydrocarbons to form carbon monoxide, carbon dioxide and hydrogen followed by further reaction to methane. The overall steam pre-reforming process may therefore be considered at least in part as a methane generation process.

15

In one embodiment, at 300 to 400°C propane, C<sub>3</sub>H<sub>8</sub>, the principal component of LPG, may be steam pre-reformed to CH<sub>4</sub>, CO, CO<sub>2</sub> and H<sub>2</sub>. The CH<sub>4</sub> in this fuel stream is then internally reformed within the fuel cell in the presence of steam during the oxidation reaction at the anode to produce a waste stream of CO<sub>2</sub> and H<sub>2</sub>O. The temperature in the fuel cell should be at least 650°C, more preferably at least 700°C to ensure substantially complete reforming of the methane. In an SOFC the temperature is likely to be at least 700°C so that complete reforming of the methane is likely to be achieved.

20

Generally, the steam pre-reforming process will be carried out such that the C<sub>2+</sub> hydrocarbon fuel is resident over the reforming catalyst for a sufficient time to ensure at least substantially complete conversion of the C<sub>2+</sub> hydrocarbons, for example to less than about 0.1% by volume in the fuel stream from the pre-reformer. This alleviates deposition of carbon on the anode when heavier hydrocarbons are reformed on the anode. However, some C<sub>2+</sub> hydrocarbons may be present in the fuel stream and preferably there is 97.5% or greater conversion of the C<sub>2+</sub> hydrocarbons in the steam pre-reforming. More preferably, there is no more than about 0.5

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- 6 -

vol% C<sub>2+</sub> hydrocarbons present in the fuel stream to the anode measured on a water basis. It is believed that reacting a fuel stream containing a substantial proportion of methane at the anode can produce higher voltages from the fuel cell than reacting hydrogen alone, and thereby improve the efficiency of the fuel cell.

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Generally the methane content of the fuel stream will be at least 25% by volume, more preferably at least about 40% by volume, even more preferably at least about 50% by volume, and most preferably at least about 60% by volume, measured on a wet basis. In a preferred embodiment, other than possibly moisture, CH<sub>4</sub> forms the major component of the fuel stream  
10 to the fuel cell.

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These levels of methane have the potential to cause excessive cooling of the fuel cell as a result of the endothermic methane internal steam reforming reaction. This problem is particularly likely to be encountered in a wholly ceramic SOFC fuel cell stack due to the low thermal conductivity of ceramic materials, but can be alleviated by incorporating metal or  
metallic components in the fuel cell stack, for example as the gas separators between individual fuel cells, to improve the thermal conductivity across the stack. Alternatively, or  
in addition, other means may be provided to alleviate excessive cooling at the fuel entry edge of each fuel cell assembly, including preheating of the fuel stream.

20

Advantageously, in the process of the invention waste heat from the fuel cell is recycled to the steam pre-reformer, which is preferably operated adiabatically. Since the pre-reformer is only required to operate at a maximum temperature of 500°C, any heat exchanger transferring the waste heat to the reformer may be of relatively simple construction and be formed of relatively  
25 low-cost materials.

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A variety of different steam reformers have been proposed and any of these may be adopted for the pre-reformer, bearing in mind the maximum operating temperature of 500°C, and the fact that the pre-reformer is effectively acting as a methane generator. The common pre-reformer catalyst is nickel-based, but may comprise, for example, platinum, rhodium, other  
precious metal, or a mixture of any of these.



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Generally, the heavy hydrocarbon fuel will pass through a desulphurising step upstream of the steam pre-reformer in order to alleviate sulphur poisoning the pre-reformer catalyst and/or the anode. Desulphurising of heavy hydrocarbon fuels is well known and will not be described further herein.

Preferably, the anode in the fuel cell comprises a nickel material, such as a nickel/zirconia cermet, which is used to catalyse the reforming reaction in the fuel cell. The fuel cell and its associated assembly can take any suitable form provided it operates at a temperature of at least 650°C to provide at least substantial conversion of the methane in the internal reforming reaction. By way of example only, several different planar SOFC components and systems, SOFCs and materials are described in our International Patent Applications PCT/AU96/00140, PCT/AU96/00594, PCT/AU98/00437, PCT/AU98/00719 and PCT/AU98/00956, the contents of which are incorporated herein by reference, including the corresponding US national phase patent 5,942,349 and patent applications 09/155061, 09/445735, 09/486501 and 09/554709, respectively. Other disclosures appear in our International patent applications PCT/AU99/01140, PCT/AU00/00630 and PCT/AU00/00631.

Generally, the fuel cell to which the fuel stream is supplied will be one of multiple fuel cells to which the fuel stream is also supplied, commonly called a fuel cell stack in the case of planar SOFCs. However, the invention also extends to the process being performed using a single fuel cell.

Commonly, steam reforming of hydrocarbons is carried out at a steam to carbon (S/C) ratio of greater than 2. In the present invention, this however would result in significant dilution of the fuel with steam and thus reduction in the fuel value. For example, for butane ( $C_4H_{10}$ ), eight volume parts of steam must be added to one volume part of fuel for an S/C ratio of 2. For diesel ( $C_{10}$ ), twenty parts of steam must be added to one part of fuel to achieve an S/C ratio of 2, with the result that there is strong fuel dilution, leading to inefficient electricity production. Preferably therefore, the S/C ratio in the pre-reformer is below 1.5, more preferably below 1.25 and most preferably below 1.

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Potential carbon deposition problems at the proposed low steam to carbon ratios are alleviated by the mild conditions (temperature no greater than 500°C) used in the pre-reformer. If pre-reforming is carried out at very low steam to carbon ratios, additional steam may be introduced to the fuel stream entering the fuel cell. Advantageously, the addition of steam may be provided by recycling some of the anode exhaust stream.

The advantages of a relatively low S/C ratio and a relatively low temperature in the pre-reformer may be seen from the thermodynamic equilibrium molar compositions of LPG and gasoline reformed at the identified maximum temperatures given in Tables 1 and 2, respectively, in which S/C is the steam to hydrocarbon carbon ratio. For example, S/C equals 1.0 is equivalent to 3.15 moles of steam for every mole of LPG or to 7.93 moles of steam for every mole of gasoline.

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25

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TABLE I

Molar Compositions of Reformed LPG (Assumed to be 85% propane and 15% butane)

Temp.	200°C						250°C					
	0.5	1.0	1.5	2.0	2.5		0.5	1.0	1.5	2.0	2.5	
S/C												
CH <sub>4</sub>	0.7133	0.4965	0.3807	0.3085	0.2591		0.7091	0.4909	0.3749	0.3028	0.2536	
C <sub>2</sub> <sup>+</sup>	4.0E-06	1.1E-06	5.7E-07	2.4E-08	3.1E-12		6.9E-06	1.9E-06	9.8E-07	8.8E-08	3.1E-12	
H <sub>2</sub> O	0.1347	0.3923	0.5306	0.6169	0.6759		0.1310	0.3845	0.5206	0.6057	0.6637	
H <sub>2</sub>	0.0041	0.0069	0.0080	0.0086	0.0090		0.0107	0.0182	0.0210	0.0224	0.0234	
CO	3.2E-05	1.3E-05	8.9E-06	6.6E-06	5.3E-06		0.0002	8.7E-05	5.8E-05	4.4E-05	3.6E-05	
CO <sub>2</sub>	0.1479	0.1042	0.0807	0.0660	0.0560		0.1490	0.1064	0.0834	0.0690	0.0592	
Total	1.0000	1.0000	1.0000	1.0000	1.0000		1.0000	1.0000	1.0000	1.0000	1.0000	
Temp.	300°C						350°C					
	0.5	1.0	1.5	2.0	2.5		0.5	1.0	1.5	2.0	2.5	
S/C												
CH <sub>4</sub>	0.7007	0.4797	0.3635	0.2918	0.2430		0.6855	0.4606	0.3448	0.2738	0.2258	
C <sub>2</sub> <sup>+</sup>	1.1E-05	3.0E-06	1.5E-06	9.1E-07	3.1E-12		1.6E-05	4.3E-06	2.1E-06	1.3E-06	3.0E-12	
H <sub>2</sub> O	0.1239	0.3691	0.5013	0.5839	0.6405		0.1127	0.3437	0.4694	0.5486	0.6031	
H <sub>2</sub>	0.0238	0.0403	0.0463	0.0493	0.0510		0.0458	0.0772	0.0880	0.0930	0.0956	
CO	0.0010	0.0004	0.0003	0.0002	0.0002		0.0040	0.0017	0.0012	0.0009	0.0008	
CO <sub>2</sub>	0.1506	0.1105	0.0886	0.0748	0.0653		0.1520	0.1168	0.0966	0.0837	0.0747	
Total	1.0000	1.0000	1.0000	1.0000	1.0000		1.0000	1.0000	1.0000	1.0000	1.0000	

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Molar Compositions of Reactant Gas Mixtures						
Temp.	250°C					
S/C	0.5	1.0	1.5	2.0	2.5	
CH <sub>4</sub>	0.7361	0.5000	0.3786	0.3046	0.2548	0.2466
C <sub>2</sub> <sup>+</sup>	7.2E-06	1.3E-06	4.9E-07	3.4E-07	1.4E-07	3.2E-12
H <sub>2</sub> O	0.0554	0.3584	0.5142	0.6092	0.6731	0.6550
H <sub>2</sub>	5.5E-11	5.1E-11	5.0E-11	4.9E-11	4.8E-11	0.0217
CO	6.5E-05	1.7E-05	1.2E-05	8.3E-06	7.1E-06	4.4E-05
CO <sub>2</sub>	0.2084	0.1416	0.1072	0.0862	0.0721	0.0767
Total	1.0000	1.0000	1.0000	1.0000	1.0000	1.0000
Temp.	350°C					
S/C	0.5	1.0	1.5	2.0	2.5	
CH <sub>4</sub>	0.7262	0.4819	0.3598	0.2861	0.2367	0.2204
C <sub>2</sub> <sup>+</sup>	2.0E-05	3.4E-06	1.6E-06	9.5E-07	3.1E-12	3.1E-12
H <sub>2</sub> O	0.0491	0.3343	0.4822	0.5724	0.6333	0.5978
H <sub>2</sub>	0.0139	0.0357	0.0424	0.0457	0.0476	0.0902
CO	0.0021	0.0006	0.0004	0.0003	0.0002	0.0009
CO <sub>2</sub>	0.2087	0.1475	0.1152	0.0955	0.0822	0.0907
Total	1.0000	1.0000	1.0000	1.0000	1.0000	1.0000

TABLE 2 cont.

TABLE 2 cont.												
Temp.	400°C						450°C					
S/C	0.5	1.0	1.5	2.0	2.5		0.5	1.0	1.5	2.0	2.5	
CH <sub>4</sub>	0.6911	0.4374	0.3158	0.2441	0.1968		0.6486	0.3986	0.2800	0.2111	0.1661	
C <sub>2</sub> <sup>+</sup>	4.0E-05	6.6E-06	2.9E-06	1.7E-06	3.0E-12		4.8E-05	8.1E-06	3.5E-06	1.9E-06	1.1E-06	
H <sub>2</sub> O	0.0415	0.2805	0.4109	0.4921	0.5477		0.0399	0.2409	0.3579	0.4328	0.4855	
H <sub>2</sub>	0.0483	0.1181	0.1380	0.1465	0.1505		0.0824	0.1826	0.2108	0.2218	0.2260	
CO	0.0247	0.0072	0.0048	0.0037	0.0031		0.0592	0.0202	0.0137	0.0107	0.0089	
CO <sub>2</sub>	0.1944	0.1568	0.1305	0.1136	0.1019		0.1699	0.1577	0.1376	0.1236	0.1135	
Total	1.0000	1.0000	1.0000	1.0000	1.0000		1.0000	1.0000	1.0000	1.0000	1.0000	
Temp.	500°C						550°C					
S/C	0.5	1.0	1.5	2.0	2.5		0.5	1.0	1.5	2.0	2.5	
CH <sub>4</sub>	0.5830	0.3461	0.2342	0.1703	0.1293		0.4996	0.2794	0.1793	0.1234	0.0885	
C <sub>2</sub> <sup>+</sup>	4.9E-05	8.9E-06	3.6E-06	1.8E-06	1.0E-06		4.5E-05	8.3E-06	3.1E-06	1.4E-06	7.3E-07	
H <sub>2</sub> O	0.0382	0.1975	0.2989	0.3671	0.4170		0.0329	0.1544	0.2402	0.3023	0.3507	
H <sub>2</sub>	0.1342	0.2597	0.2951	0.3072	0.3104		0.2032	0.3455	0.3842	0.3949	0.3944	
CO	0.1135	0.0481	0.0336	0.0266	0.0222		0.1793	0.0957	0.0694	0.0554	0.0462	
CO <sub>2</sub>	0.1311	0.1486	0.1382	0.1288	0.1211		0.0849	0.1250	0.1269	0.1240	0.1202	
Total	1.0000	1.0000	1.0000	1.0000	1.0000		1.0000	1.0000	1.0000	1.0000	1.0000	

TABLE 2 cont.

Temp.	600°C					700°C				
	0.5	1.0	1.5	2.0	2.5	1.0	1.5	2.0	2.5	
S/C										
CH <sub>4</sub>	0.4156	0.2048	0.1201	0.0755	0.0492	0.0831	0.0302	0.0133	0.0066	
C <sub>2</sub> <sup>+</sup>	4.1E-05	6.3E-06	2.0E-06	7.9E-07	2.3E-12	2.1E-06	9.1E-12	2.1E-12	2.2E-12	
H <sub>2</sub> O	0.0227	0.1142	0.1875	0.2464	0.2962	0.0515	0.1197	0.1886	0.2520	
H <sub>2</sub>	0.2776	0.4336	0.4696	0.4740	0.4659	0.5744	0.5872	0.5620	0.5286	
CO	0.2409	0.1568	0.1188	0.0954	0.0788	0.2594	0.2058	0.1621	0.1291	
CO <sub>2</sub>	0.0432	0.0906	0.1040	0.1087	0.1099	0.0316	0.0571	0.0740	0.0837	
Total	1.0000	1.0000	1.0000	1.0000	1.0000	1.0000	1.0000	1.0000	1.0000	
Temp.	750°C					700°C (cont.)				
	3.0	3.5	1.0	1.5	2.0	2.5	3.0	3.5		
S/C										
CH <sub>4</sub>	0.0036	0.0021	0.0488	0.0112	0.0042	0.0019	0.0010	0.0006		
C <sub>2</sub> <sup>+</sup>	2.2E-12	2.3E-12	1.1E-06	1.3E-10	2.0E-12	2.1E-12	2.2E-12	2.3E-12		
H <sub>2</sub> O	0.3085	0.3583	0.0322	0.1078	0.1851	0.2532	0.3120	0.3630		
H <sub>2</sub>	0.4947	0.4629	0.6157	0.6095	0.5699	0.5294	0.4922	0.4588		
CO	0.1045	0.0861	0.2866	0.2268	0.1768	0.1406	0.1142	0.0944		
CO <sub>2</sub>	0.0887	0.0906	0.0167	0.0447	0.0640	0.0749	0.0806	0.0832		
Total	1.0000	1.0000	1.0000	1.0000	1.00000	1.0000	1.0000	1.0000	1.0000	



30 The fuel stream, optionally with steam from the anode outlet side of the fuel cell stack added to it, is introduced to the anode side of the stack 14. The fuel cells operate at a temperature of at least 700°C and when the fuel stream contacts the nickel/zirconia cermet anodes the

methane in the fuel stream is steam reformed to carbon monoxide and hydrogen.

At the same time oxygen, in the form of air, is supplied to the cathode side of the fuel cell stack and, when the fuel cell stack is short-circuited through an external load (not shown), the fuel oxidises at the anodes resulting in a CO<sub>2</sub> and H<sub>2</sub>O waste stream.

### Examples

#### Example 1:

Using a steam pre-reforming process to produce high levels of methane, propane was subjected to steam reforming using a commercially available catalyst at two different maximum temperatures of 378°C and 328°C and at a variety of S/C ratios. The catalyst was a nickel-based pre-reformer catalyst C11 - PR supplied by United Catalysts Corporation. The measured conversions given in Table 3 illustrate that the results predicted by the thermodynamic calculations given above can be at least approached. The steam reforming was conducted adiabatically in a microreactor system with on-line analysis of the gas compositions. It is to be noted that the experimental set-up was not ideal, and it is believed that this accounts for the incomplete conversions. All conversions would usually be expected to be at least 97.5% complete with appropriate residence times over the catalyst to ensure this. All percentages are vol% and are given on a dry basis.

TABLE 3

Temp (°C)	Steam/Carbon ratio	Steam/Gas ratio	H <sub>2</sub> %	CO %	CH <sub>4</sub> %	CO <sub>2</sub> %	C <sub>3</sub> H <sub>8</sub> %	Conversion of Propane
378	1.37	4.11	34	2.4	48.9	14.2	0.6	97.5
328	1.37	4.11	15.4	2	62.7	15	4.8	84.6
328	1.01	3.03	24.4	3.8	56.5	12.4	3	89.2
328	0.75	2.25	21.4	1.6	61.1	13.1	3.1	89.0
328	0.5	1.5	17.3	1.6	64.6	12.9	3.6	88.2

Examples 2 – 4

Further experiments were conducted on the production of a methane-rich fuel for internal reforming on SOFC anodes from LPG available in Victoria, Australia ("Victorian LPG") in a prototype steam pre-reformer. The composition of Victorian LPG varies from 100% propane to a mix of propane, butane and/or ethane. The composition of the LPG used during this work was 93% propane and 7% ethane.

The experiments were carried out in a prototype fuel processor with 0.6L of the aforementioned commercial pre-reforming catalyst C11-PR. Operating conditions were chosen to have higher methane content in the reformed gas than is normally generated from conventional pre-reforming of propane and ethane, with a sufficient residence time over the catalyst to give 100% conversion of the propane and ethane. The results of the experiments are given below, together with the operating conditions. It should be noted that the results are given on a dry basis.

15

Example 2:

Temp: 353-380°C, Steam/Carbon=1.5.

CH <sub>4</sub> :	48.3
H <sub>2</sub> :	30.6
CO <sub>2</sub> :	21.1
CO:	0.0
C <sub>3</sub> H <sub>8</sub> :	0.0
C <sub>2</sub> H <sub>6</sub> :	0.0

25

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Example 3:

Temp: 361-393°C, Steam/Carbon=1.0.

CH<sub>4</sub>: 54.9  
5 H<sub>2</sub>: 25.0  
CO<sub>2</sub>: 20.1  
CO: 0.0  
C<sub>3</sub>H<sub>8</sub>: 0.0  
C<sub>2</sub>H<sub>6</sub>: 0.0

10

Example 4:

Temp: 367-402°C, Steam/Carbon=0.75.

CH<sub>4</sub>: 57.9  
15 H<sub>2</sub>: 22.5  
CO<sub>2</sub>: 19.5  
CO: 0.0  
C<sub>3</sub>H<sub>8</sub>: 0.0  
C<sub>2</sub>H<sub>6</sub>: 0.0

20

The above examples confirm that it is possible to steam reform heavy hydrocarbons (C<sub>3+</sub>) at relatively low temperatures and achieve both at least 20 vol% methane on a wet basis in the resultant fuel stream and at least 97.5% conversion of the heavy hydrocarbons.

25 Example 5

In this example a gas comprising approximately 28 vol% CH<sub>4</sub>, 22 vol% H<sub>2</sub>, 12 vol% CO<sub>2</sub>, 1 vol% CO and 37 vol% H<sub>2</sub>O was supplied to a fuel cell for internal reforming and electricity production.

30

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As may be seen from Table 1 (molar% and vol% are substantially the same) the gas composition simulated LPG (85 vol% propane, 15 vol% butane) which has been steam reformed at 450°C and an S/C ratio of 1.5.

5 The single fuel cell was a standard planar SOFC with a 50 micron thick cathode layer of strontium doped lanthanum manganite (LSM) on one side of a 120 micron thick electrolyte layer of dense 3 mol% yttria stabilised zirconia (3YSZ) and a 50 micron thick anode layer of 50% Ni-3YSZ cermet on the other side of the electrolyte layer. The fuel cell was mounted in a closed ceramic housing with a fuel side inlet and outlet and fuel gas flow channels on the  
10 anode side and an air side inlet and outlet and air flow channels on the cathode side. A seal was provided in the housing around the fuel cell to prevent mixing of the fuel gas and air. Electrical current was taken off using Ni mesh in contact with the anode, between the anode and the anode side of the housing, and a Pt mesh in contact with the cathode, between the cathode and the cathode side of the housing.

15

The fuel cell was heated to 930°C and the fuel gas was supplied to the anode side inlet while air was supplied to the cathode side inlet. At this temperature, the methane in the fuel gas was internally steam reformed on the anode to hydrogen and oxides of carbon which were then oxidised on the anode as a result of the fuel reaction. The resultant voltage output from the  
20 fuel cell was measured for a period of about 55 hours and is shown in Figure 2.

Over extended periods of time substantially greater than 55 hours, it is possible that some carbon deposition from the methane internal reforming reaction may detrimentally affect the fuel cell performance. A variety of approaches have been put forward to alleviate this, of  
25 which the addition of promoters to Ni steam reforming catalysts may be particularly appropriate in the present invention, should it prove necessary.

Such promoters include alkali metals (eg Na, K) and alkali earth metals (Mg, Ca, Sr, Ba) (J.R. Rostrup-Nielson in Catalysis Science and Technology, Volume 5, Springer, Berlin, 1984, p1).  
30 Alkali additions are not advisable for SOFC anodes, but alkali earth additions (Mg, Ca, Sr, Ba) and ceria additions have been demonstrated for suppression of carbon deposition {P.Singh

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et al. US Patent 4 894 297 (1990); V.D. Belyaev et al, Applied Catalysis A, 133, p47(1995)).

Addition of Mo-oxide has also been shown to be effective for carbon deposition suppression (R.H. Cunningham et al, Solid Oxide Fuel Cells 5, Proceedings Volume 97-40, The Electrochemical Society, Pennington, NJ, 1997, p973).

5

Those skilled in the art will appreciate that the invention described herein is susceptible to variations and modifications other than those specifically described. It is to be understood that the invention includes all such variations and modifications which fall within its spirit and scope. The invention also includes all of the steps, features, compositions and compounds referred to or indicated in this specification, individually or collectively, and any and all combinations of any two or more of said steps, features, compositions and compounds.

10

The reference to any prior art in this specification is not, and should not be taken as, an acknowledgment or any form of suggestion that that prior art forms part of the common general knowledge in Australia or elsewhere.

15

Throughout this specification and the claims which follow, unless the context requires otherwise, the word "comprise", and variations such as "comprises" and "comprising", will be understood to imply the inclusion of a stated integer or step or group of integers or steps but not the exclusion of any other integer or step or group of integers or steps.

20

**CLAIMS:**

1. A process for producing electricity in a fuel cell which comprises reacting a higher carbon ( $C_{2+}$ ) hydrocarbon fuel with steam in a steam pre-reformer at a temperature in the pre-reformer of no greater than  $500^{\circ}\text{C}$  to produce a fuel stream including hydrogen and no less than about 20% by volume methane measured on a wet basis, and supplying the fuel stream and an oxidant to a high temperature fuel cell in which the methane is reformed and electricity is produced by reacting the fuel stream at an anode of the fuel cell and reacting the oxidant at a cathode of the fuel cell.
2. A process according to claim 1 in which the fuel stream includes no less than about 25% by volume methane measured on a wet basis.
3. A process according to claim 2 in which the fuel stream includes no less than about 40% by volume methane, even more preferably no less than about 50% by volume methane, and most preferably no less than about 60% by volume methane, measured on a wet basis.
4. A process according to any one of claims 1 to 3 in which the temperature in the steam pre-reformer is no more than about  $450^{\circ}\text{C}$ , more preferably in a range of about  $250^{\circ}\text{C}$  to  $450^{\circ}\text{C}$  and most preferably in a range of about  $300^{\circ}\text{C}$  to  $400^{\circ}\text{C}$ .
5. A process according to any one of claims 1 to 4 in which the reaction of the fuel with steam in the methane generator is performed adiabatically.
6. A process according to any one of claims 1 to 5 in which the steam to carbon ratio in the methane generator is no more than 1.5, more preferably no more than 1.25 and most preferably no more than 1.0.
7. A process according to claim 6 in which additional steam is introduced to the fuel stream downstream of the steam pre-reformer.

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8. A process according to claim 7 in which the additional steam is recycled from an anode exhaust stream.
- 5 9. A process according to any one of claims 1 to 8 in which the fuel is a  $C_{3+}$  hydrocarbon fuel.
10. A process according to claim 9 in which the fuel is selected from the group consisting of LPG, gasoline (petrol) and diesel, kerosene, fuel oil and jet fuel.
- 10 11. A process according to any one of claims 1 to 10 in which the fuel is selected from the group consisting of ethane, propane, butane, LPG, gasoline (petrol), diesel, kerosene, fuel oil, jet oil, naphtha and mixtures of two or more of these.
- 15 12. A process according to any one of claims 1 to 11 in which the reaction at the anode of the fuel cell is performed at a temperature of at least  $700^{\circ}\text{C}$ .
13. A process according to any one of claims 1 to 12 in which waste heat from the fuel cell is recycled to the steam pre-reformer.
- 20 14. A process according to any one of claims 1 to 12 in which the reaction in the steam pre-reformer results in the conversion of at least 97.5% of the higher carbon ( $C_{2+}$ ) hydrocarbon fuel.
- 25 15. A process according to claim 14 which results in at least substantially complete conversion of the higher carbon ( $C_{2+}$ ) hydrocarbon fuel.



## AMENDED CLAIMS

[received by the International Bureau on 14 December 2000 (14.12.00);  
original claim 10 amended; other claims unchanged (1 page)]

8. A process according to claim 7 in which the additional steam is recycled from an anode exhaust stream.
9. A process according to any one of claims 1 to 8 in which the fuel is a C<sub>3+</sub> hydrocarbon fuel.
10. A process according to claim 9 in which the fuel is selected from the group consisting of LPG, gasoline (petrol) and diesel.
11. A process according to any one of claims 1 to 10 in which the fuel is selected from the group consisting of ethane, propane, butane, LPG, gasoline (petrol), diesel, kerosene, fuel oil, jet oil, naphtha and mixtures of two or more of these.
12. A process according to any one of claims 1 to 11 in which the reaction at the anode of the fuel cell is performed at a temperature of at least 700°C.
13. A process according to any one of claims 1 to 12 in which waste heat from the fuel cell is recycled to the steam pre-reformer.
14. A process according to any one of claims 1 to 12 in which the reaction in the steam pre-reformer results in the conversion of at least 97.5% of the higher carbon (C<sub>2+</sub>) hydrocarbon fuel.
15. A process according to claim 14 which results in at least substantially complete conversion of the higher carbon (C<sub>2+</sub>) hydrocarbon fuel.

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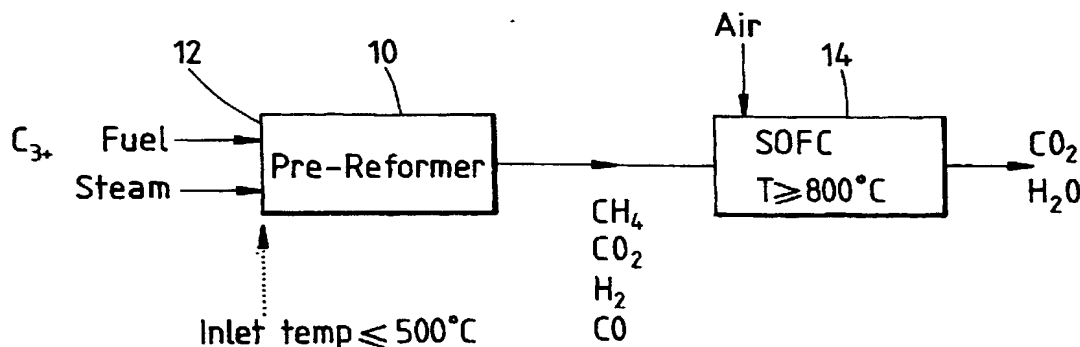
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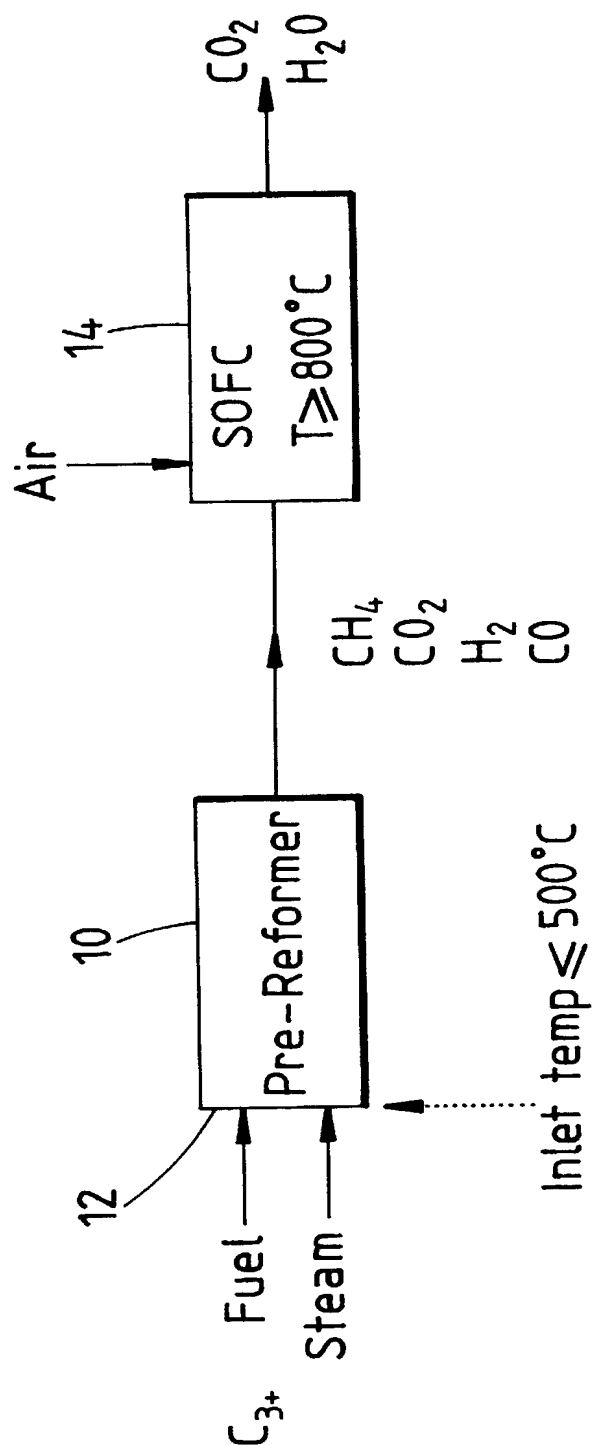
(54) Title: FUEL CELL SYSTEM



(57) Abstract: A process for producing electricity in a fuel cell (14) which comprises reacting a higher carbon (C<sub>2+</sub>) hydrocarbon fuel with steam in a steam pre-reformer (10) at a temperature in the pre-reformer of no greater than 500 °C to produce a fuel stream including hydrogen and no less than about 20 % by volume methane measured on a wet basis, and supplying the fuel stream and an oxidant to a high temperature fuel cell (14) in which the methane is reformed and electricity is produced by reacting the fuel stream at an anode of the fuel cell and reacting the oxidant at a cathode of the fuel cell.

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FIG 1

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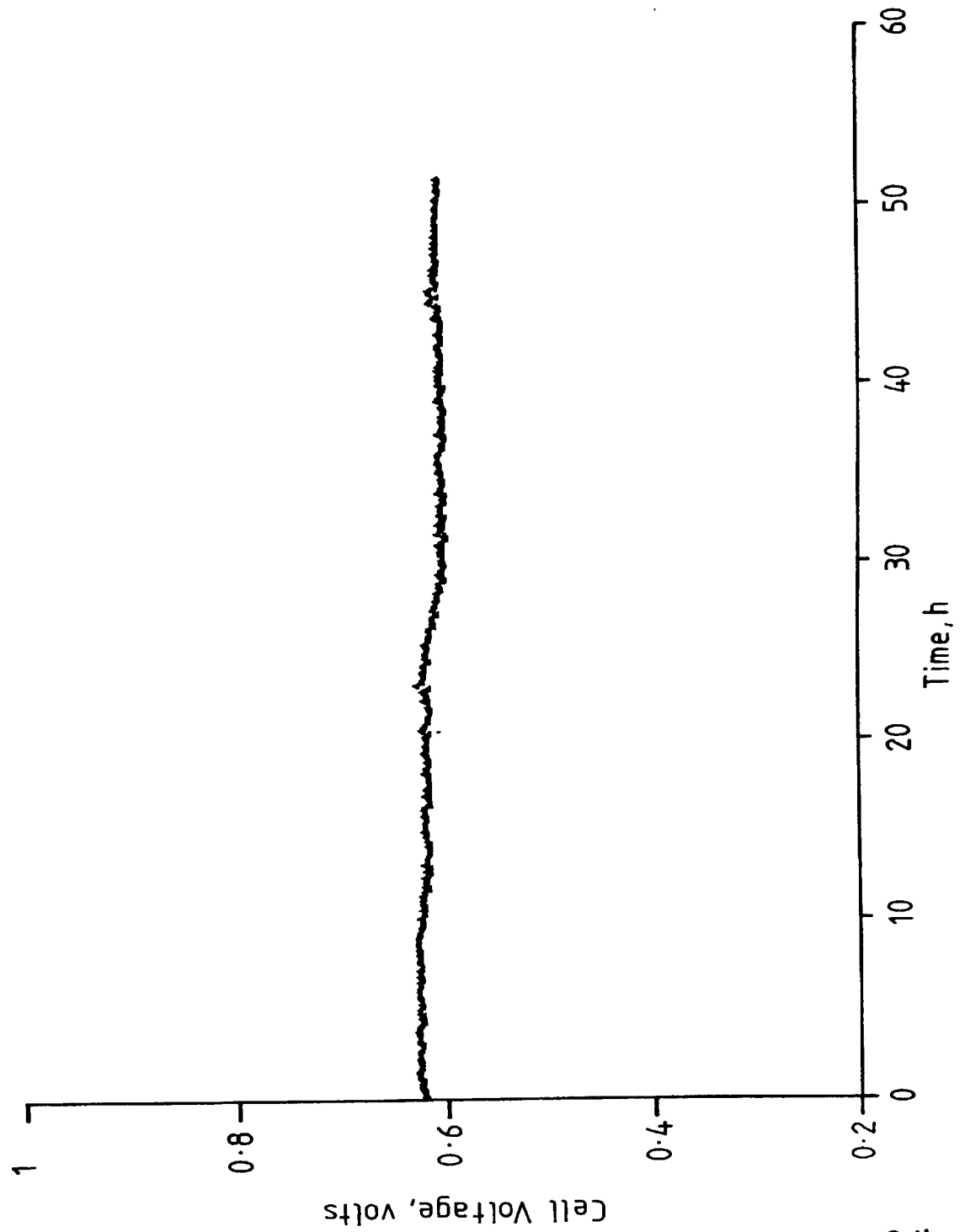


FIG 2

## DECLARATION FOR PATENT APPLICATION

As a below-named inventor, I hereby declare that:

My residence, post office address and citizenship are as stated below next to my name,

I believe I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural names are listed below) of the subject matter which is claimed and for which a patent is sought on the invention entitled Fuel cell system

the specification of which

[x] International Patent Application No. PCT/AU00/00974 filed 16 August 2000

☐ is attached hereto

(Check  
one)

☐ was filed on \_\_\_\_\_ as  
Application  
Serial No. \_\_\_\_\_  
and was  
amended on \_\_\_\_\_  
(if  
applicable)

I hereby state that I have reviewed and understand the contents of the above-identified specification, including the claims, as amended by any amendment referred to above.

I acknowledge the duty to disclose to the Patent Office all information known to me to be material to patentability as defined in 37 C.F.R. 1.56.

I hereby claim foreign priority benefits under Title 35, United States Code, § 119 of any foreign application(s) for patent or inventor's certificate listed below and have also identified below any foreign application for patent or inventor's certificate having a filing date before that of the application on which priority is claimed:

Prior Foreign Application(s)			Priority Claimed	
<u>PQ2234/99</u> (Number)	<u>Australia</u> (Country)	<u>16 August 1999</u> (Day/Month/Year	<input checked="" type="checkbox"/>	<input type="checkbox"/>
		Filed)	Yes	No
<u>PCT/AU00/00974</u> (Number)	<u>PCT</u> (Country)	<u>16 August 2000</u> (Day/Month/Year	<input checked="" type="checkbox"/>	<input type="checkbox"/>
		Filed)	Yes	No
<u>                    </u> (Number)	<u>                    </u> (Country)	<u>                    </u> (Day/Month/Year	<input type="checkbox"/>	<input type="checkbox"/>
		Filed)	Yes	No

[illegible]

I hereby claim the benefit under Title 35, United States Code, § 120 of any United States application(s) listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States application in the manner provided by the first paragraph of Title 35, United States Code, § 112, I acknowledge the duty to disclose to the Patent Office all information known to me to be material to patentability as defined in 37 C.F.R. 1.56 which occurred between the filing date of the prior application and the national or PCT international filing date of this application:

(Application Serial No.)	(Filing Date)	(Status) (patented, pending, abandoned)
(Application Serial No.)	(Filing Date)	(Status) (patented, pending, abandoned)
(Application Serial No.)	(Filing Date)	(Status) (patented, pending, abandoned)

## DECLARATION FOR PATENT APPLICATION

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I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Title 18, United States Code, § 1001 and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

1-00

Full name of sole or first inventor:	* <u>Karl FOGER</u>
Inventor's signature:	<i>Karl Foger</i>
Date:	<u>02/05/02</u>
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Citizenship:	* Australian <span style="float: right;">DUX</span>
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2-00

Full name of second joint inventor, if any:	* <u>Khaliq AHMED</u>
Inventor's signature:	<i>Khaliq</i>
Date:	<u>18. 4. 2002</u>
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Full name of third joint inventor, if any:	*
Inventor's signature:	
Date:	
Residence:	*
Citizenship:	*
Post Office Address:	*

Full name of fourth joint inventor, if any:	*
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DECLARATION FOR PATENT APPLICATION